Glaze It Forward



by Diana Pancioli

WHY USE CONE 6 REDUCTION?

First of all Cone 6 saves money. Some say it costs less by almost a third compared to Cone 10. Secondly, it helps preserve the kiln; Cone 6 is much easier on kilns than Cone 10. Thirdly, less gas used means less carbon emissions produced and a greener planet for polar bears and humans. Lastly, a reduction atmosphere makes beautiful glazes, whatever the temperature.

The development of cone 6 Reduction glazes was originally an attempt to insure that our expensive new gas kiln would last a long time. It has. Extending its life was my major goal ten years ago; saving fuel and reducing emissions were secondary, until now.

Having fired cone 10 R for many years, I also thought that it might be fun to add something to the literature by developing reduction glazes for a lower temperature. After several years of firing a handful of glazes at 6R, I applied for a summer grant to develop a better palette. I tested hundreds of glazes—anything I could find that was written for mid-range. The 22 glazes chosen for Glaze Forward are the product of that summer's research. (Thank you Eastern Michigan University.)

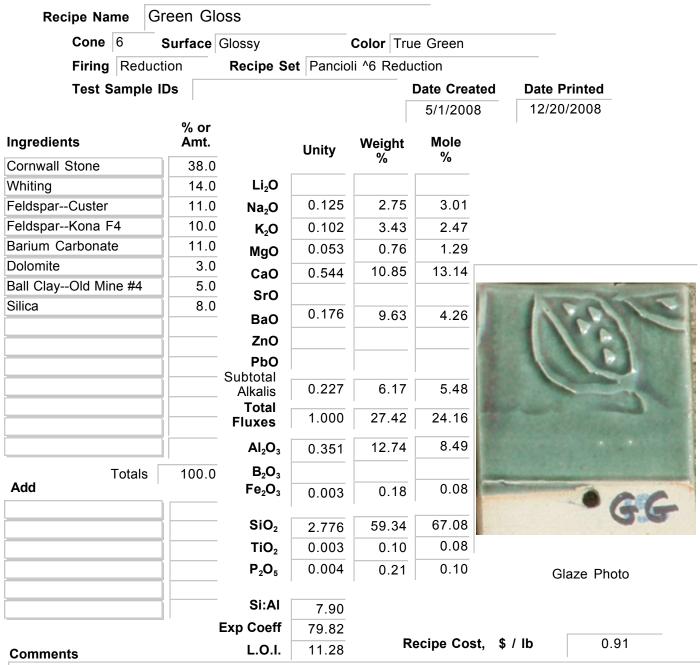
My first goal was to lower the temperature of traditional Cone 10R glazes-- celadon, temmoku, iron saturate, shino, copper red, etc. I revised some favorite formulae to the new lower temperature. The remainder were selected from many tests; I hoped to provide a range of colors, surfaces, and bases that would satisfy many tastes and encourage experimentation by others.

I have provided a list of the sources of these glazes. You will recognize some of them from cone 10. I renamed all the glazes according to their surface qualities so that (a) they would not be mistaken for their cone 10 versions and (b) their names would give my students a clue about each glaze's color and appearance.

I sent some glazes to the Alfred Analytical Testing Laboratory for leach testing. The results are reported in the notes at the bottom of each glaze page. Some testing still needs to be done.

John Hesselberth has been generous in allowing me to use a version of his software "GlazeMaster" to present the glaze recipes, their photos, and their chemical analyses for you. (Thank you John.)

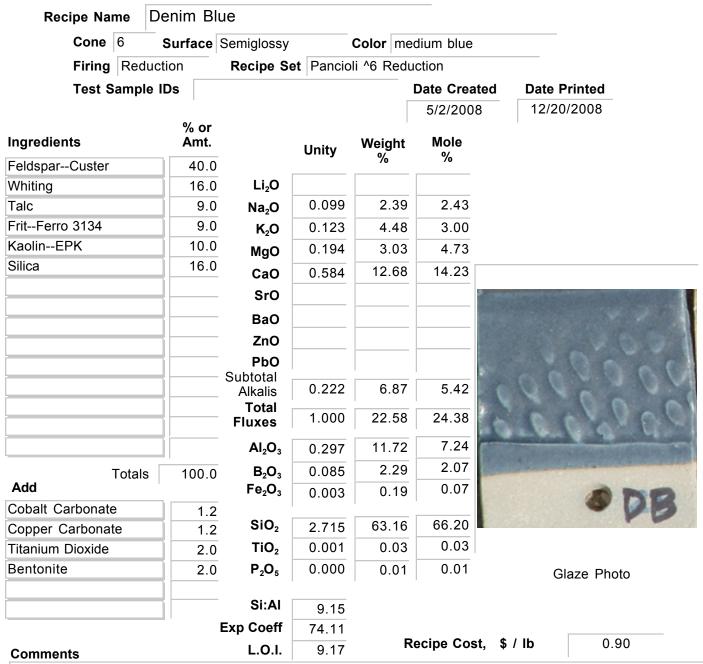
Diana Pancioli Glaze Forward January, 2009



Derived from Cone 10 "Dimig Green"

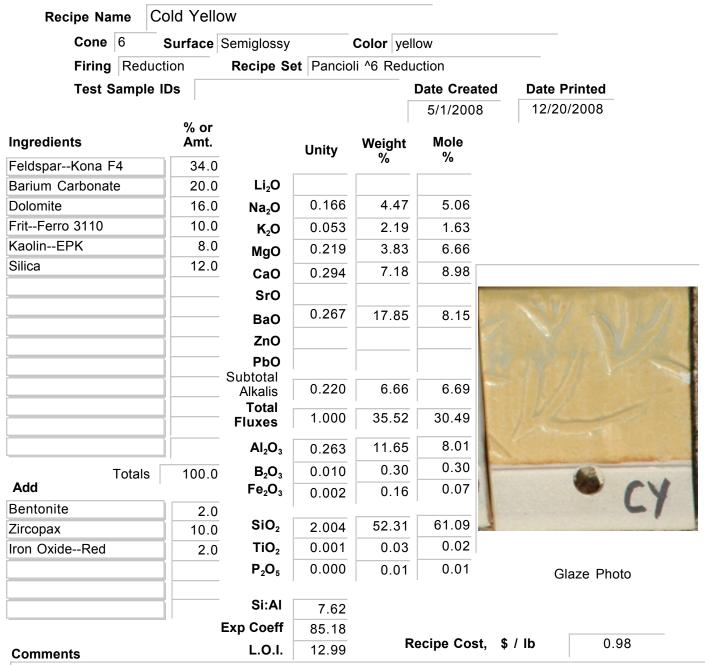
This beautiful glossy green stays green unless the reduction is too heavy. Sometimes a pink blush will highlight one side of the piece. This glaze crazes. (Probably more flint will fix it). It is a stronger green with 5% copper but also leaches more. Becasue of the high copper content and the barium it is not useful as a liner. Strontium substitution for the barium doesn't work. It helps fix the crazing but the color changes to a muddy pink. At 4% copper carbonate it leached 23.2 mg/l copper

At 5% it leached 43.1 mg/l copper and 14.2 barium. Thin to medium application



This is a Val Cushing glaze; the master has done it again. This glaze is probably stable enough to pass a leach test, which means you can use it at these colorant levels on the interiors of vessels for food use.

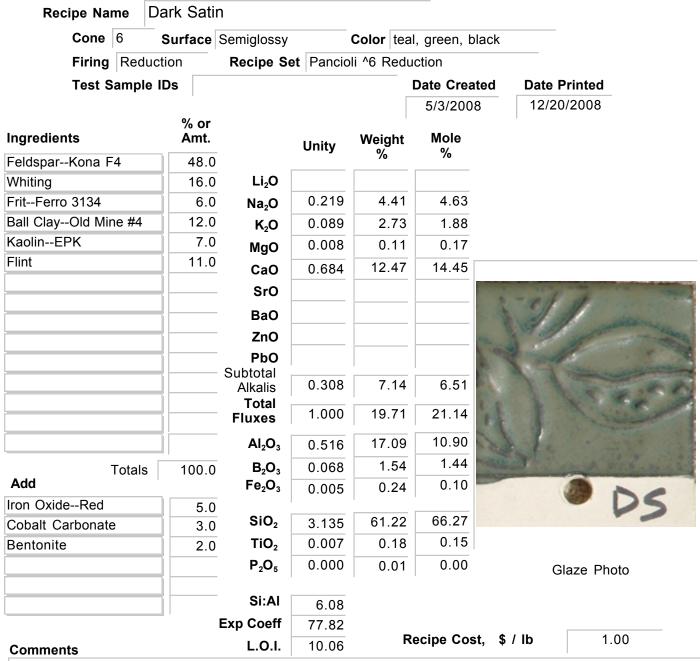
Leach test results not yet completed.



Derived from Cone 10 "Alfred Yellow"

This cold yellow glaze goes to pale blue where thick and warm brown where thin. Strontium does not work as a substitute for barium in this glaze. The yellow color is completely lost with strontium.

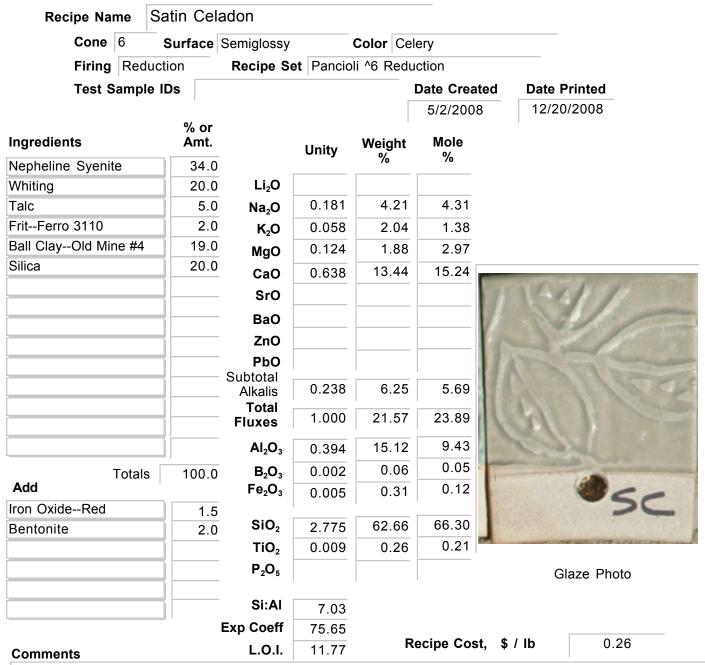
This is not a liner glaze. It leaches barium at 7.25 mg/l.



Derived from E. Cooper's "Matt Iron"

A textured satin with variegated color. The attractive teal/green/black color mix will change to a smooth satin black with more dark colorants but may also then be in danger of leaching. A dark glaze should contain no more than 2 to 3 percent cobalt if it is to be used as a liner. This glaze is more colorful on an iron-bearing clay body.

At 6% cobalt (twice as much as the recipe calls for) it leached 0.1 mg/l of cobalt.

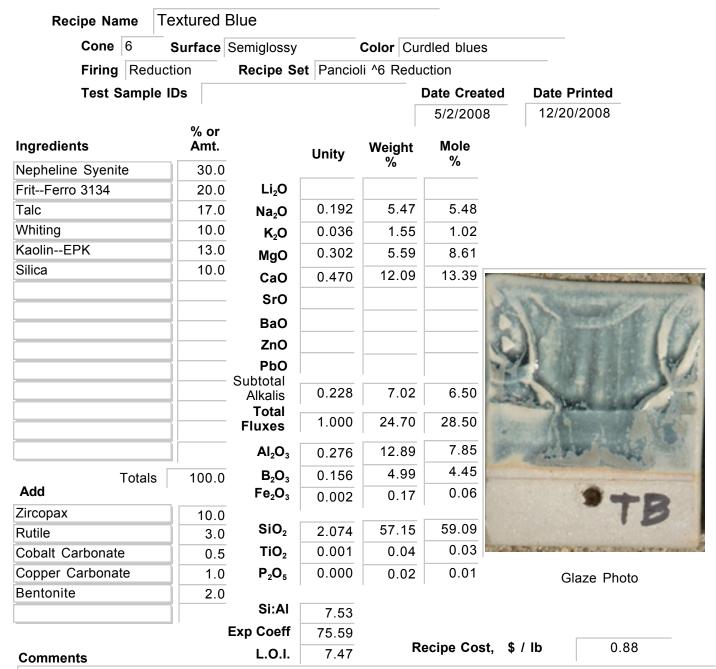


Derived from E. Cooper's "Clear Pale Olive"

A lovely satin celadon (more celery colored, not blue). The addition of more silica will make it shinier and move toward correcting the crazing (which is only visible if you stain it). But I don't like the glaze as well with the silica addition. I added a little frit to give it a slight sheen.

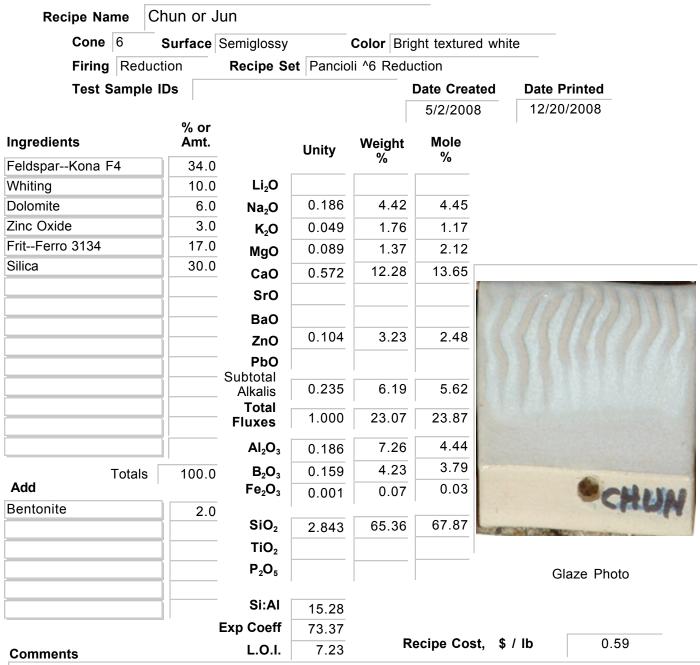
This glaze settled out more than the others despite the addition of bentonite so I added a small amount of Epsom salts to keep it in suspension

Not as nice on porcelain.



This is Marcia Selsor's Waxy White base with a number of colorants added. This variation is derived from a 50/50 color blend with rutile incorporated in the base for texture. There are other nice combinations with rutile but this one remains a favorite. Gloes glossy on interiors and breaks beautifully over textures.

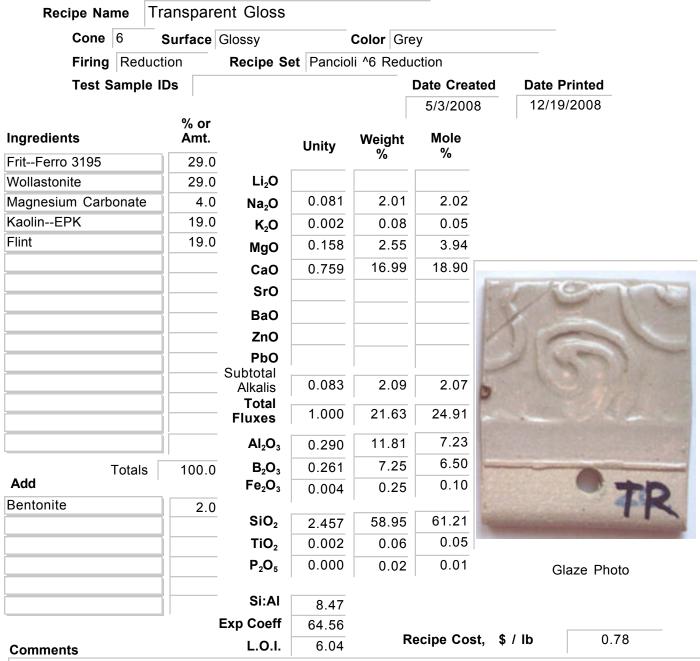
Medium application.



MarciaSelsor's "Jun", unchanged.

This is a beautiful bright marshmellow-like white textured semi-gloss. I tried adjusting this formula but it didn't come out as well as the original. (You might want to try to take out the zinc.) The difficulty is that to make this glaze more stable one has to destroy an important quality--its ability to run and hold itself in beautiful fat drips. (Don't use near the foot.)

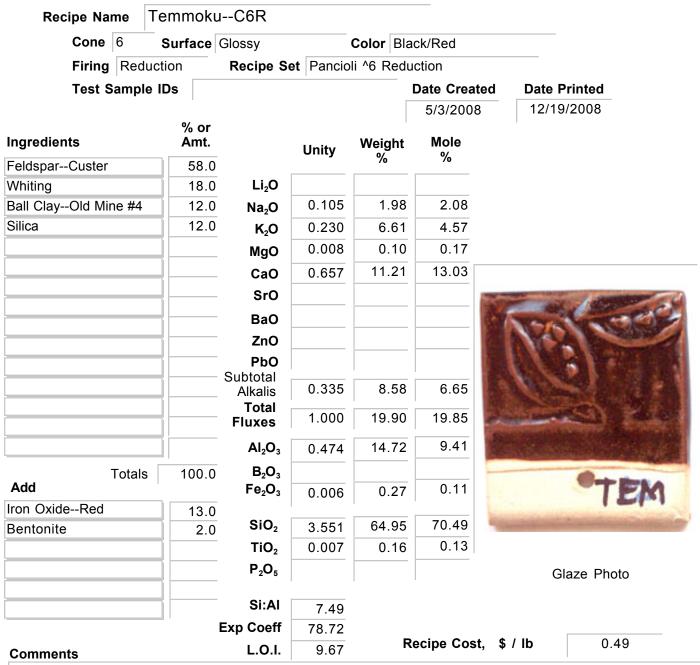
I accept that it might not last as long with the dishwasher use (although after 3 years of use I can't see any changes). Application depends on where you want the glaze to form drips. (It is good lightly tinted too, with small amounts of iron or nickel.)



An adjustment of Pete Pinnell's "1214"

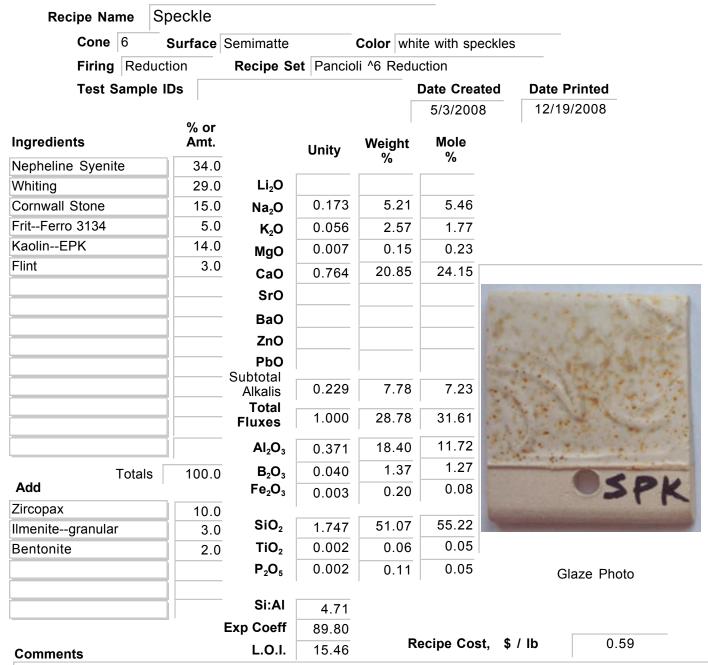
This transparent fits quite well on the clay body I used, which shrinks 13%. This glaze may be used as a liner. A small addition of iron will convert it to a grey/green celadon.

Thin application.



An adjustment of Hamer's "Temmoku"

This cone 6 Temmoku has 13% iron, a lot more than a Temmoku at Cone 10. At cone 6 reduction Temmokus seem to need more than the usual 8% required at Cone 10 to give the same black/red color break. At 11% iron this formula makes a handsome orange glaze with darker brown coloration in the depths.

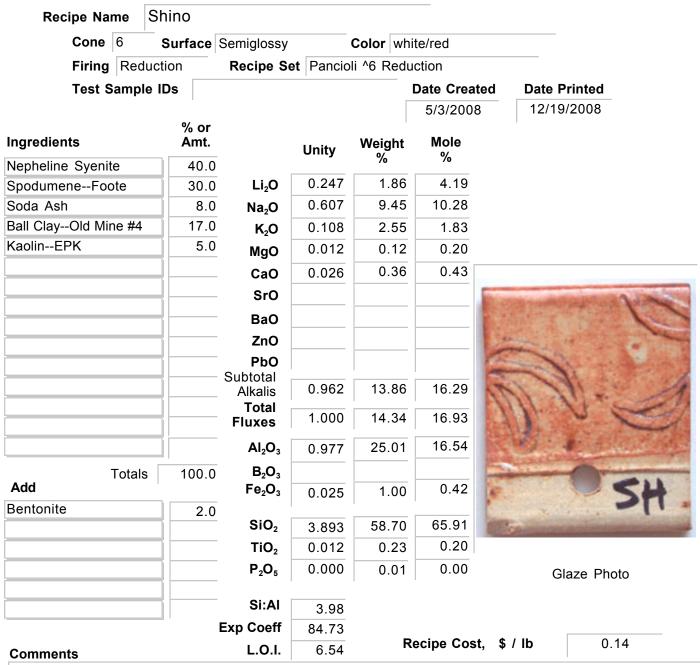


Derived for E. cooper's "Cream Matt Speckle"

A smooth off-white semi-matte with orangy speckles.

The matte is not useful as a liner.

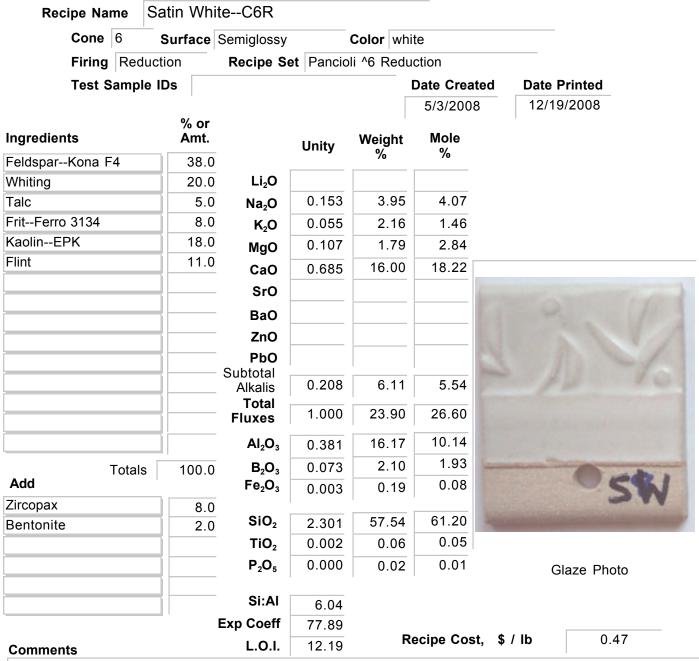
Apply thin to medium.



Derived from a cone 10 glaze "Porcelain Shino"

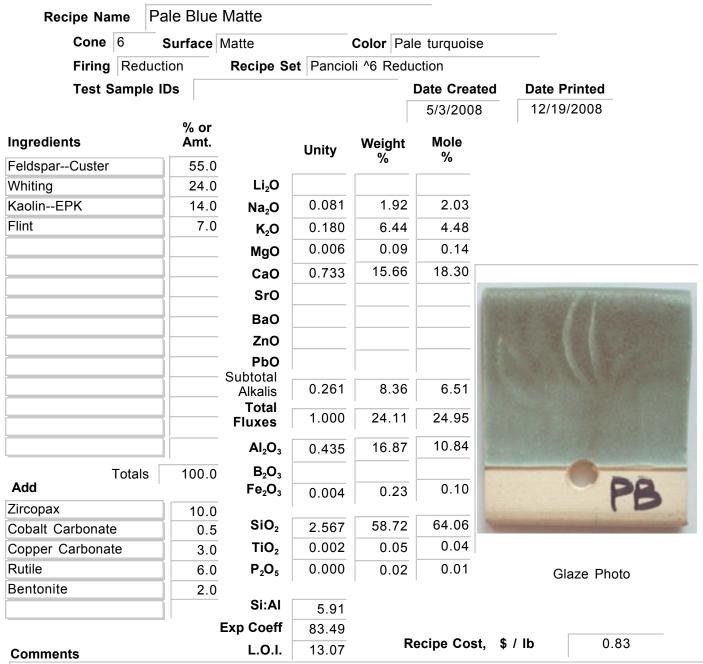
Shinos are temperamental. This one needs strong reduction and is more effective on a dark iron body. It is rather bland on white stoneware. It is capable of carbon trapping. It also crazes. This glaze does not work well thickly applied.

Application should be thin to medium.



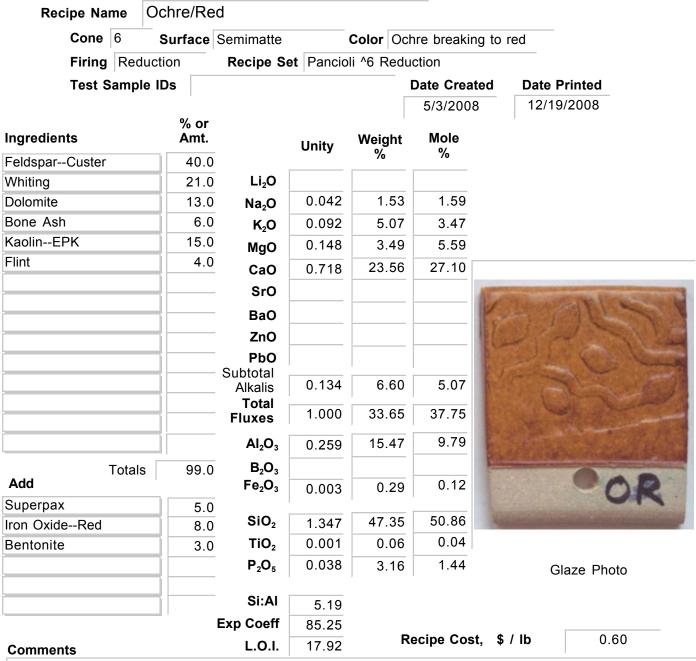
Derived from E. Cooper's "Cool Matt Blue"

This is a cool white satin. If you want it more matte, remove 2% frit. If you want it a little shinier, add 2% frit. Can be used as a liner glaze.



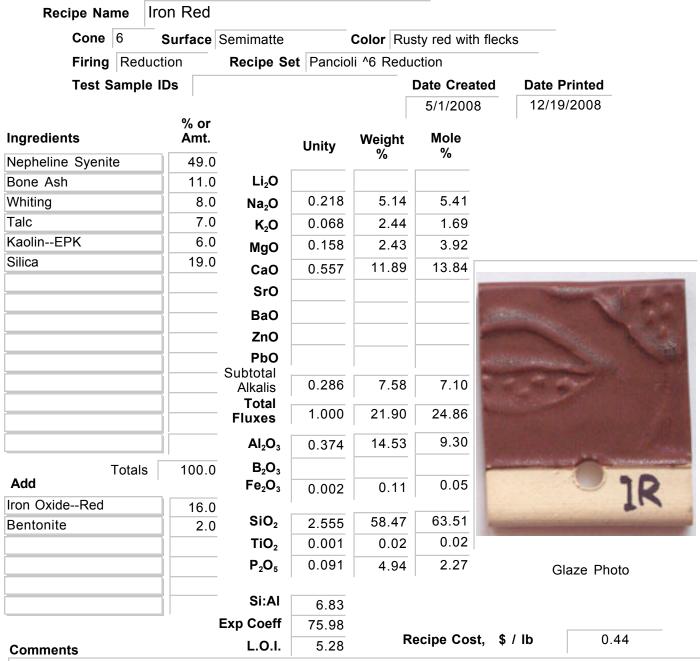
Derived from Rick Malgrem "Bronze/Green Matte"

A light blue which breaks to a darker grey in deep places. Too matte to be used as a liner.



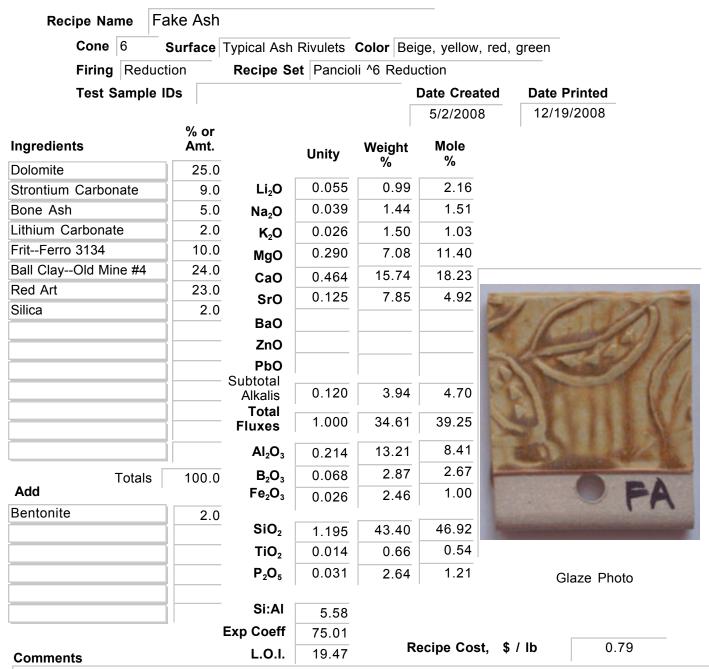
Derived from Hayden "Cinnamon Red"

This glaze, as indicated by its name, gives a nice color shift from ochre to iron red. It is low in silica so it is not a candidate for a liner glaze for food use.



Derived from Cone 10 "Oharata Red"

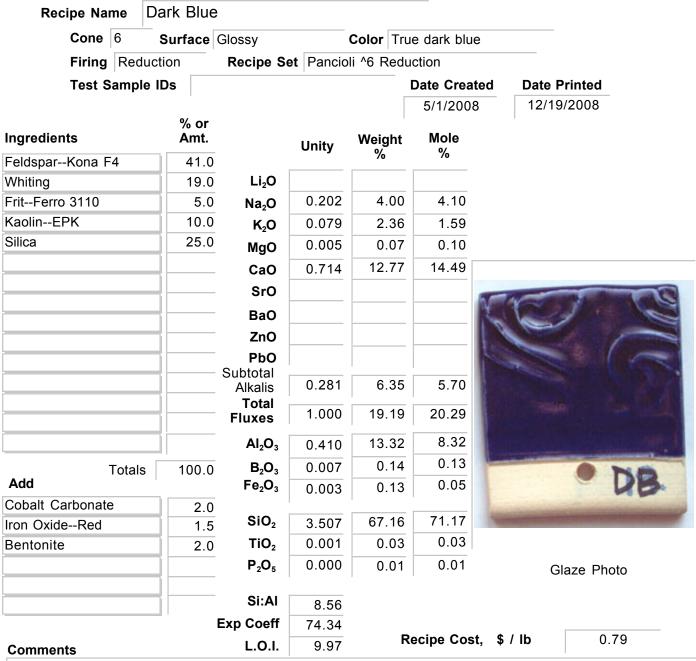
This iron red satin glaze is low in silica so probably it is not good for use as a liner for food containers. but it has an interesting ability to form smallsparkly crystals spaced out in the glaze. The addition of 10% flint will make it a shiny two toned iron saturate and will make it useful as a liner.



An adjustment of "Andie's Fake Ash"

This is a beautifully variegated fake ash glaze. It is a brighter yellow on porcelain with hints of green where thicker and terra cotta colored where thin. it is not stable because it is low in silica but to alter it would change the ash effect. While it might not meet strict requirements of stability, I use it anyway (having substituted strontium for barium). It is lovely on plates and bowls. If you wish to be more careful in your use of it you could put it instead on vases, spoon jars, dry storage jars, and other non wet food items. It is more beautiful with strontium as a substitute (one of the few I tested that is).

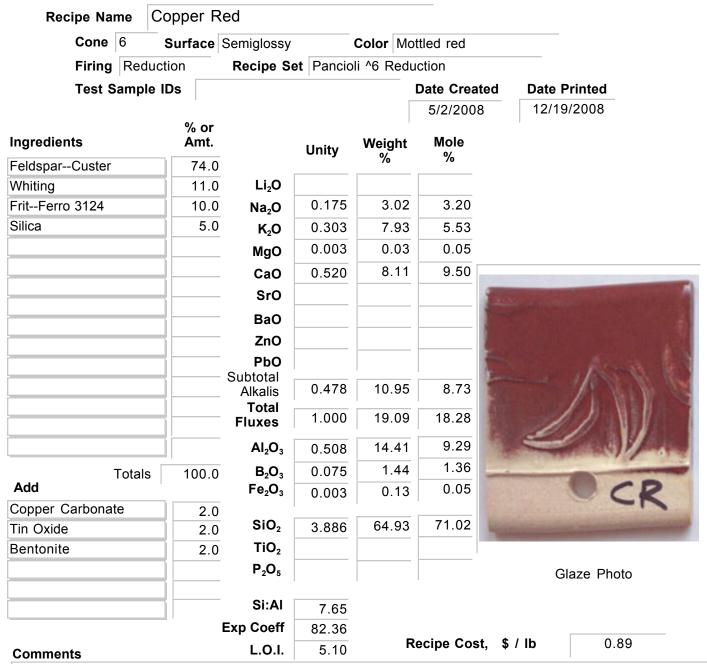
When made with barium it leached 3.28 mg/l



Source: A mystery

This is a glossy dark blue. The addition of a small amount of iron keeps the blue from moving toward violet. It may be used as a liner glaze if the cobalt is kept at or under 2 percent. Brighter on porcelain.

No cobalt leaching was detected with 2% CoCO3 in the formula.

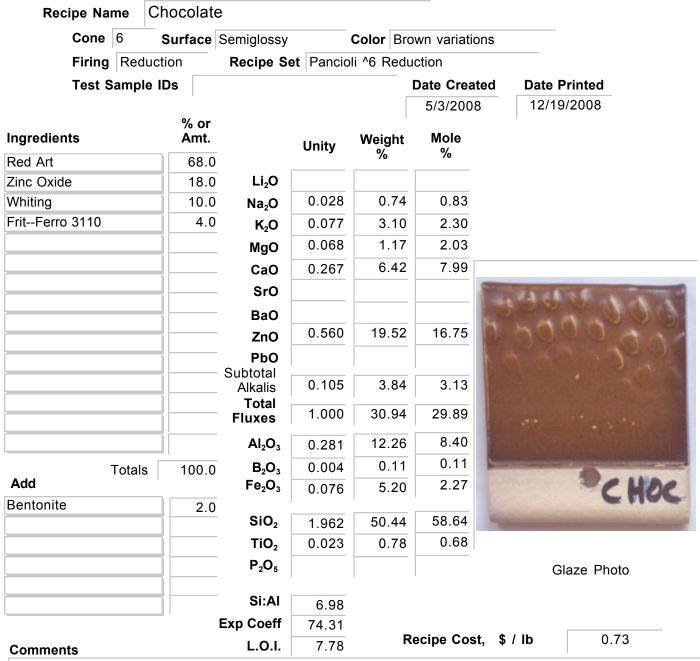


An adjustment of Pete Pinnell's "Cranberry"

This is a dependable bright copper red. It sometimes gives an attractive mottled surface.

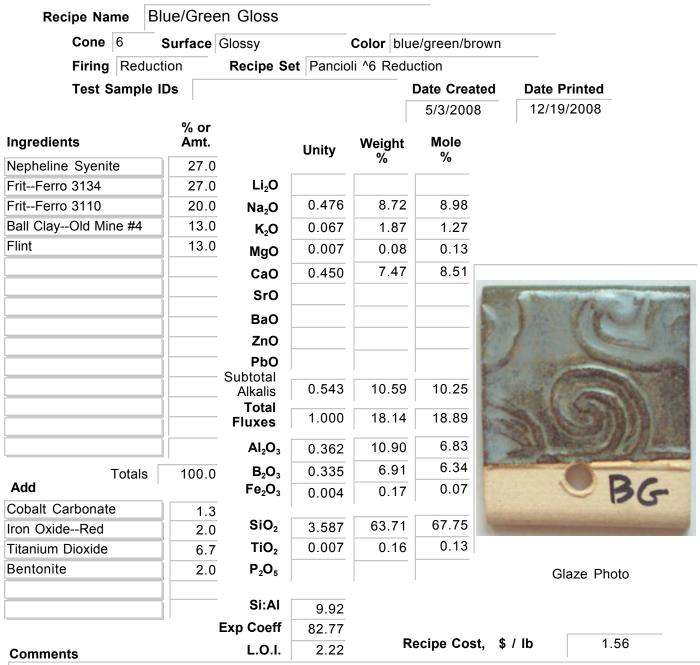
It is not suitable as a liner glaze.

It leaches 12.2 mg/l copper



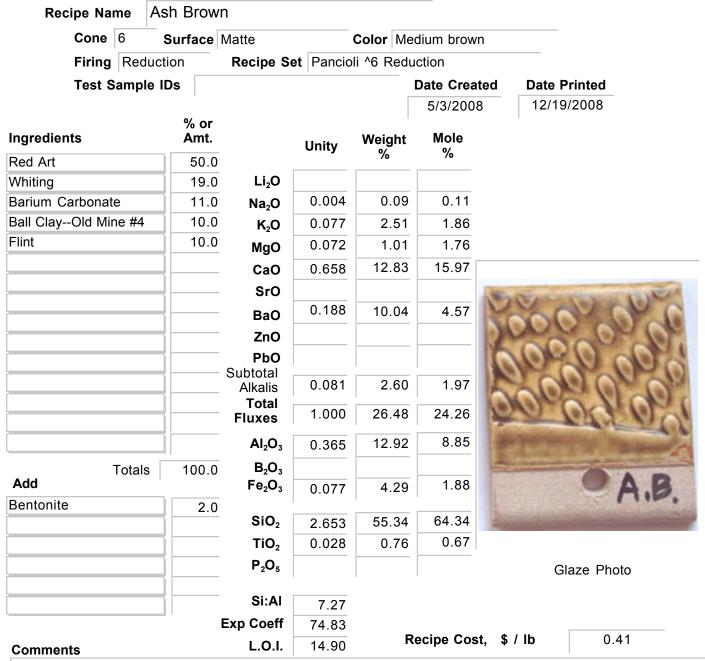
Derived from E. Cooper's "Brown Matte"

Although zinc is supposed to disappear in reduction, this glaze doesn't work without it. Chocolate looks like a slip glaze similar to old Albany slip glazes. I woul try using it under trailing, brush work, etc. The color varies subtly depending on where the flame touches it.



Derived from E. Cooper's "Medium Blue".

This glaze has nice color shifts from light blue to medium green where thicker, to dark brown where thin. May be stable. I added a small amount of Epsom salts to keep it in good suspension because the large quantity of frit in this glaze can cause it to settle badly, despite the addition of bentonite.



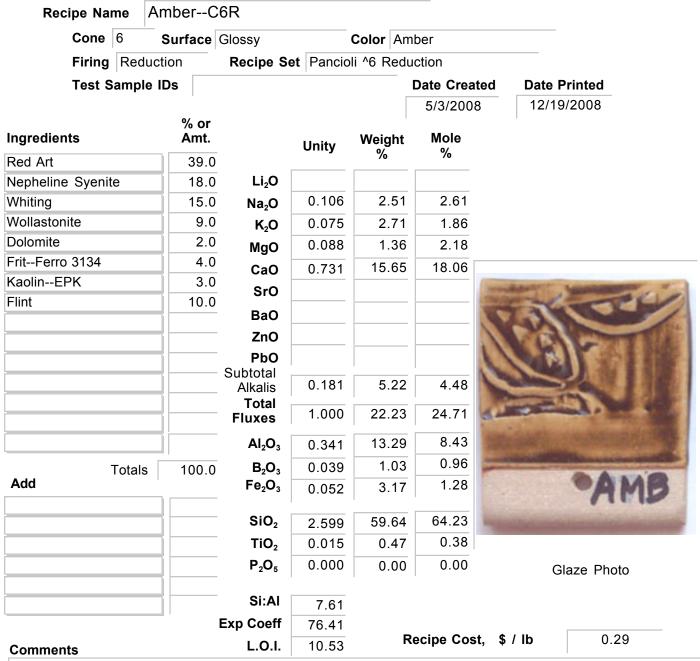
Derived from E. Cooper's "Broken Yellow Brown"

The substitution of strontium in this glaze was less than successful. The original glaze has a surface complexity which was lost with the substitution.

This is not a liner glaze.

Thin to medium application

Leach test results not vet in.



Derived from cone 10 "Amber Celadon"

I tested a number of variations of this recipe; I hope to simplify it, but the resulting changes produced a much less interesting glaze. You could try adding a percent of iron if you want it a little darker.